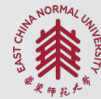




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ECNU

Improved mechanistic modeling on reproducing particle-bound mercury in the marine atmosphere

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Outline

- Introduction
- Methods and materials
- Results and discussion
- Conclusion

Hg

The Periodic Table of the Elements
by Robert Cawston - version 1.4

atomic mass
or most stable mass number
1st ionization energy
in kJ/mol

chemical symbol
name
electron configuration

atomic number
electronegativity

alkali metals
alkaline metals
other metals
transition metals
lanthanoids
actinoids

metalloids
nonmetals
halogens
noble gases
unknown elements
radioactive elements have masses in parenthesis

oxidation states
most common are bold

group 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18

period 1
2
3
4
5
6
7

electron configuration blocks
s
p
d
f

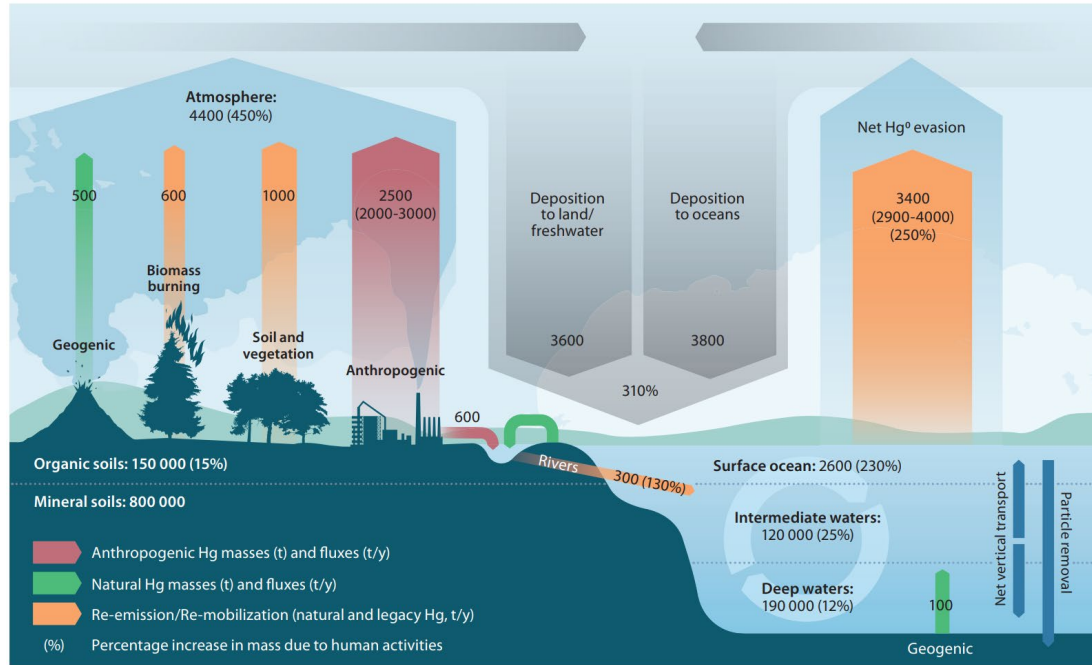
notes
• as of yet, elements 113-118 have no official name designated by the IUPAC.
• 1 kJ/mol = 96.485 eV.
• all elements are implied to have an oxidation state of zero.

200.59	80	+4
1007.1	2.00	+2
Hg		+1
Mercury		
[Xe] 4f ¹⁴ 5d ¹⁰ 6s ²		

Hg⁰ Hg²⁺ Hg^P

Ubiquitous on the planet!

Hg cycling and health impacts



UN Environment (2019)



Modeling Hg^P formation in global models

Statistics-based schemes

$$K_p = \frac{[\text{PBM}]}{[\text{PM}_{2.5}][\text{GOM}]}$$

$$\log(1/K_p) = a + \frac{b}{T}$$

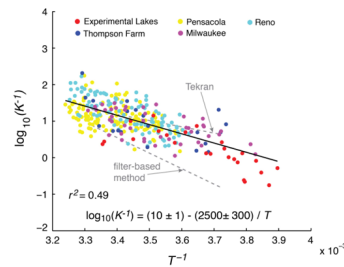
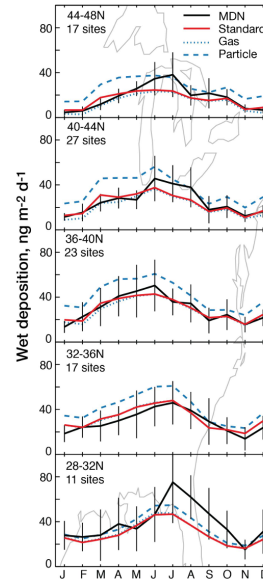
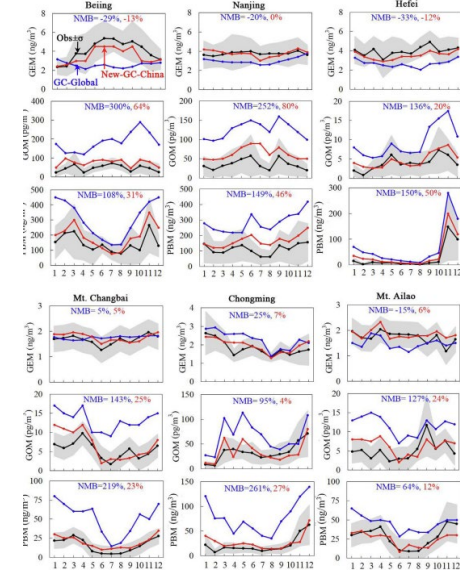


Table 2 – Hg^{II} gas-particle partitioning in New-GC-China.

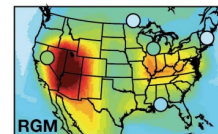
Region	RH	OA	Hg ^{II} gas-particle partitioning	Sites
Eastern coastal grids below 10 m a.s.l.	/	/	$\log(1/K_p) = -(1200 \pm 170)/T + (5 \pm 1)$	Chongming
Inland areas	RH < 60%	/	$\log(1/K_p) = -(2700 \pm 280)/T + (10 \pm 1)$	Beijing Nanjing
Inland areas	60% ≤ RH ≤ 80%	/	$\log(1/K_p) = -(2100 \pm 230)/T + (8 \pm 1)$	
Inland areas	RH > 80%	< 8 μg/m ³	$\log(K_p) = 0.25 \times \log[\text{OA}] - 1.00$	
Inland areas	RH > 80%	≥ 8 μg/m ³	$\log(K_p) = -1.14 \times \log[\text{OA}] + 0.25$	



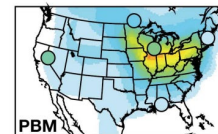
Good performance over the continents



100 175 250 ppq



0 1.5 3 ppq

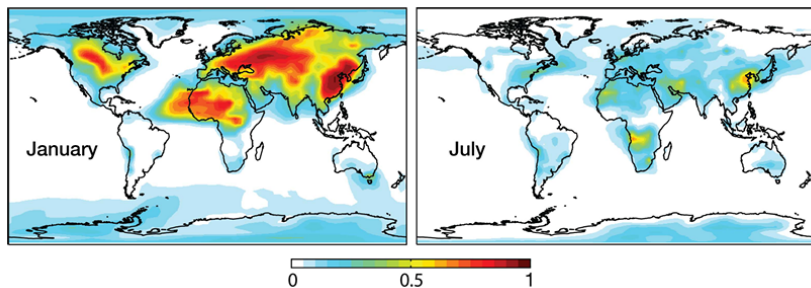


0 1.5 3 ppq

(Amos et al., 2012; Liu et al., 2022)

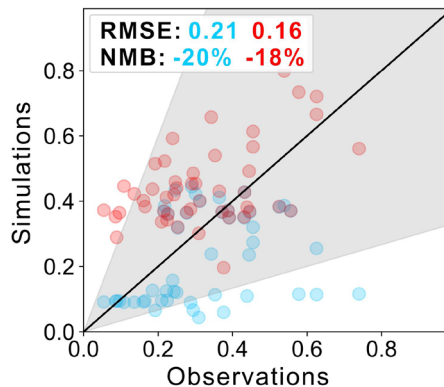
Modeling Hg^P formation in global models

□ Statistics-based schemes

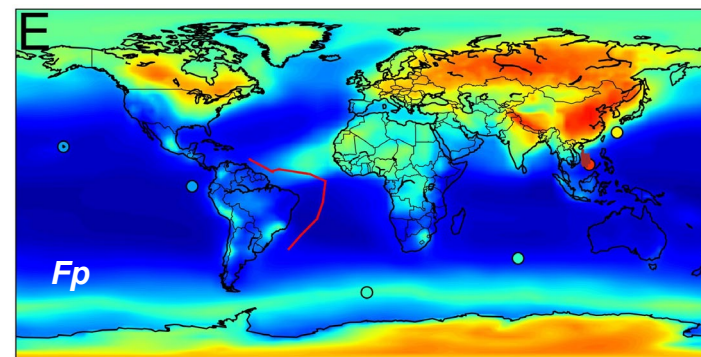
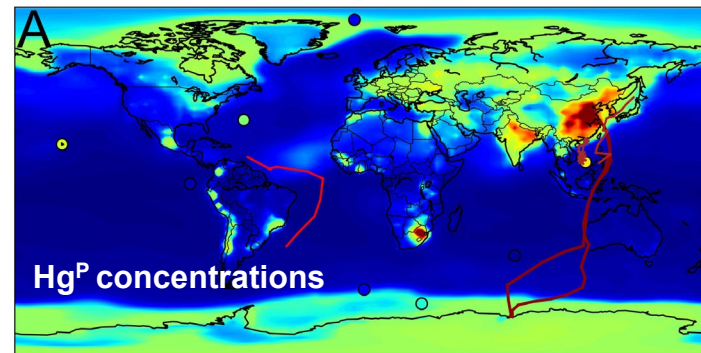


$\text{Hg}^{\text{II}}/(\text{Hg}^{\text{II}}+\text{Hg}^{\text{P}})$

Underestimate!

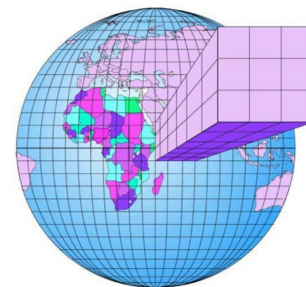
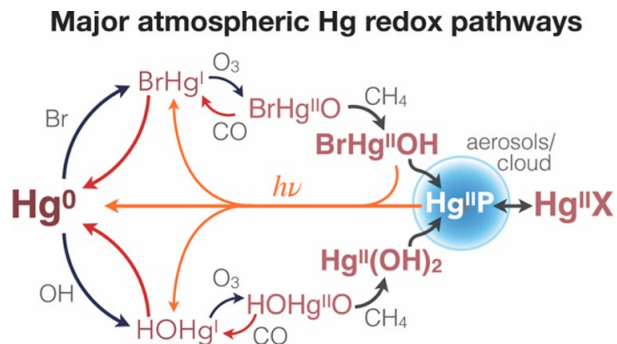


How about the performance over the ocean?



$F_p = \text{Hg}^{\text{P}}/(\text{Hg}^{\text{II}}+\text{Hg}^{\text{P}})$

- GEOS-Chem (v14.1.1) Hg modeling
(*Shah et al., 2021; Feinberg et al., 2022*)



- Anthropogenic inventory with long-term Hg emissions
Streets et al., 2019: 2000-2015
Scaling based on CO_2 emissions (ODIAC): 2016-2023
- Long-term simulations (2008–2023) at latitude $2^\circ \times$ longitude
 2.5° horizontal resolution

□ Process-based schemes

Hygroscopic growth of sea salt particles

$$r = r_{dry} \frac{4}{3.7} \left(\frac{2-S}{1-S} \right)^{1/3}$$

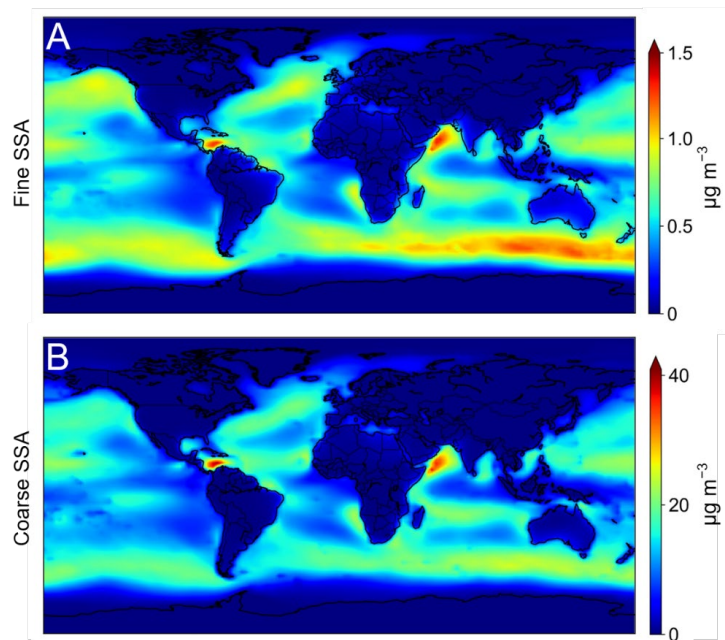
$$\rho_{wet} = (\rho r_{dry}^3 + \rho_{H_2O}(r^3 - r_{dry}^3))/r^3$$

Adsorption of Hg^{II} on the particles

$$k_{mt} = \frac{4\pi r^2}{L} \left(\frac{r}{D_g} + \frac{4}{v\alpha} \right)^{-1} N \quad \text{Mass transfer rate}$$

$$J = Lk_{mt} \left(C - \frac{C_a}{H} \right) \quad \text{Concentration gradients}$$

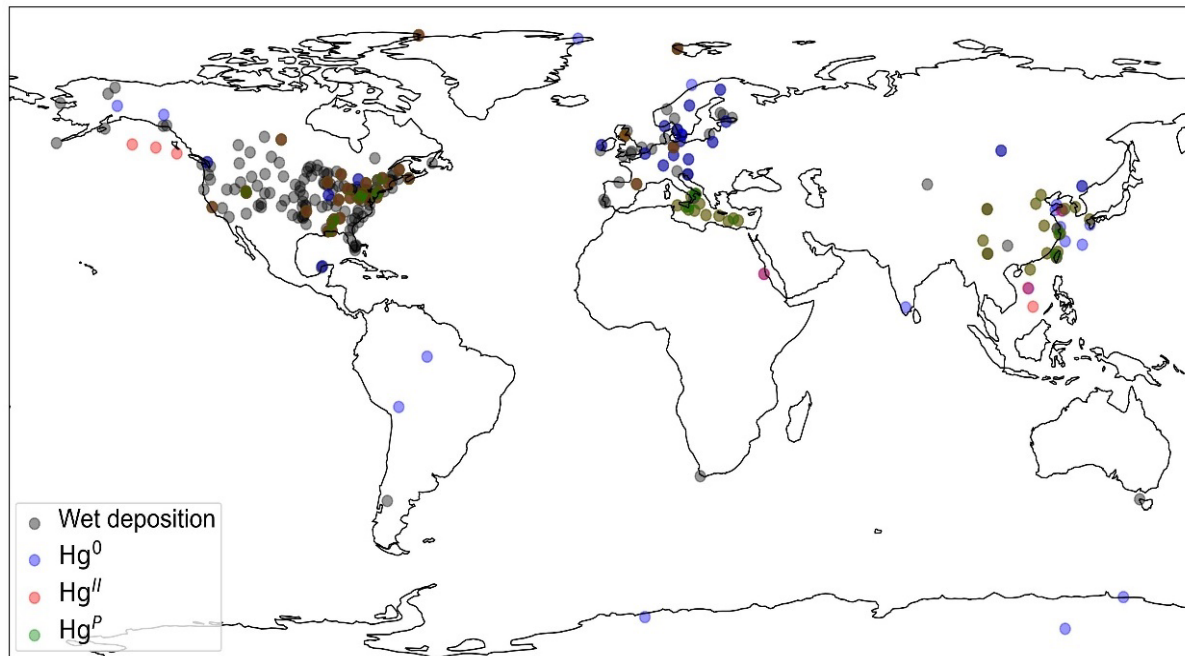
$$\frac{dC_{a,i}}{dt} = \frac{J_i}{L_i} - v_{d,i} \frac{C_{a,i}}{Z}$$



Nine bins of dry radii ranging from 0.01 to 8 μm were established in this study in accordance with Fan et al. (2011): **0.01–0.2, 0.2–0.4, 0.4–0.6, 0.6–0.8, 0.8–1, 1–2, 2–4, 4–6, and 6–8 μm**

Atmospheric Hg observations

□ Inland/Coastal observations



- CAPMoN
- AMNet
- EMEP
- GMOS
- MDN
- Literature data

Atmospheric Hg observations

□ Marine observations from offshore islands and cruises

- **Five ship cruises** across the ocean:

- South Atlantic Ocean

- West Pacific Ocean

- Weddell Sea

- North Atlantic Ocean

- Southern Ocean

- **Five offshore islands:**

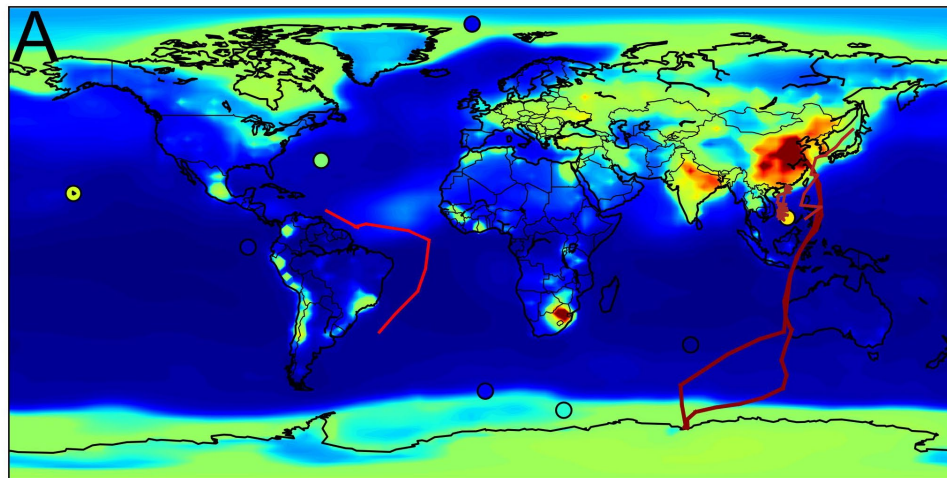
- Hawaii islands

- Bermuda

- Amsterdam Island

- The Galápagos islands

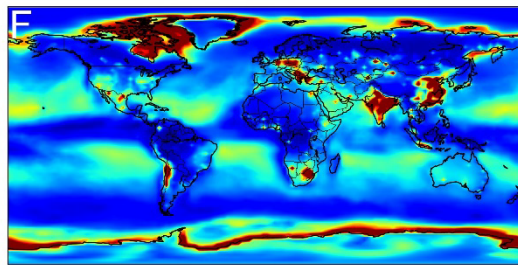
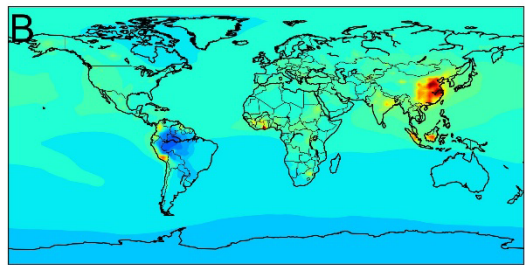
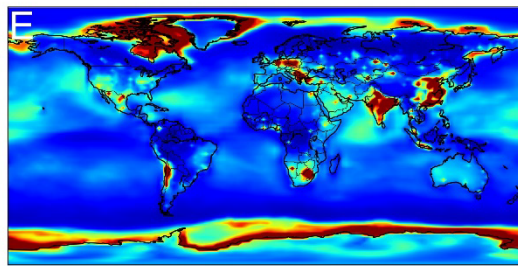
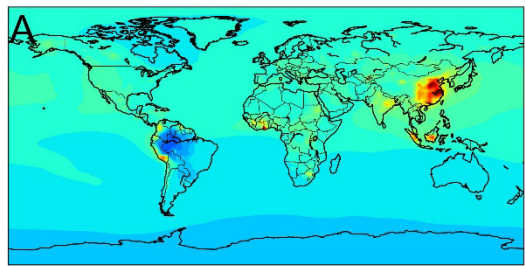
- Taiping Island



Correction of atmospheric Hg^{II} concentrations

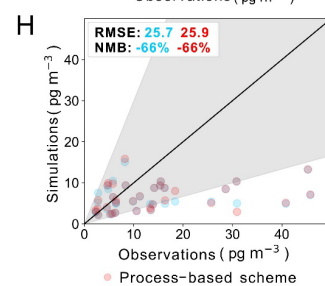
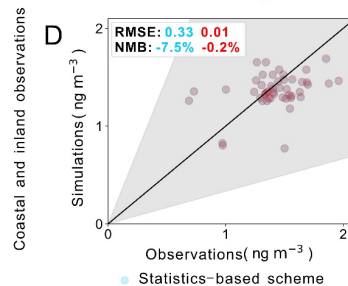
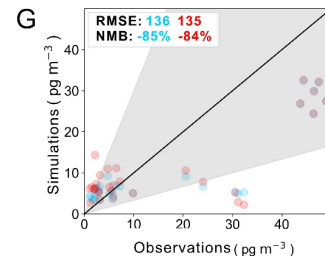
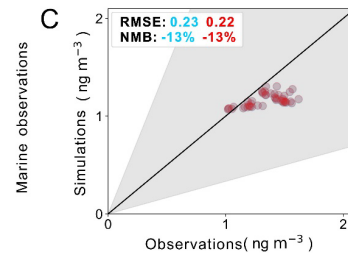
- ❑ Numerous studies have reported the underestimation of Hg^{II} concentrations measured by KCl-coated denuders from the widely-used Tekran speciation systems, and the factors of underestimation ranged from **1.3 to 13**
- ❑ A median value of **2.4** was used to correct the observations of Hg^{II} concentrations in this study, and the median value has been observed for the measuring efficiency of cation exchange membranes (CEM) over KCl-coated denuders on collecting HgCl₂, which is the main component of Hg^{II} in the atmosphere

Model performance on Hg^0 and Hg^{II} concentrations

Atmospheric Hg^0 concentrationsAtmospheric Hg^{II} concentrations

0 1 2 3 ng m^{-3}

0 10 20 pg m^{-3}



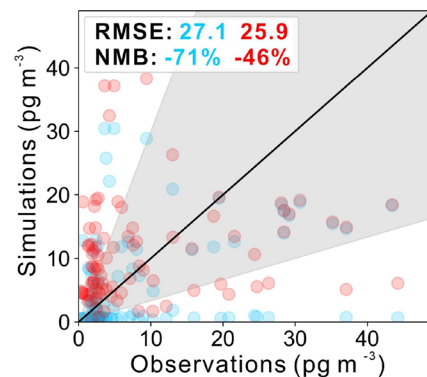
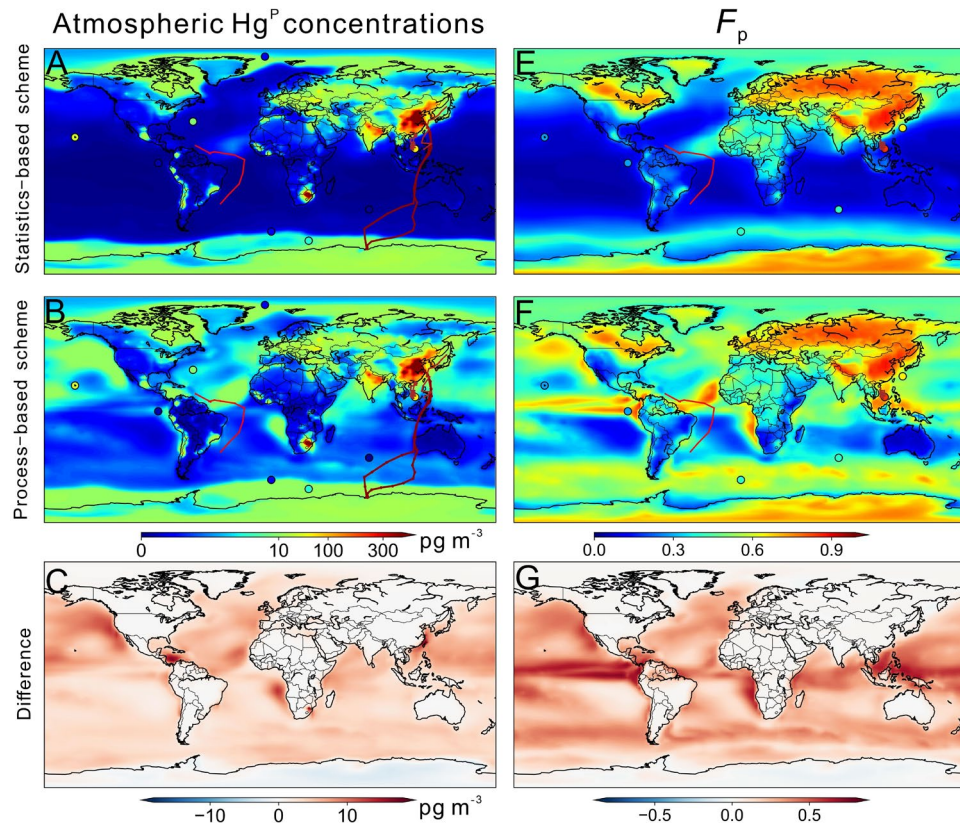
● Statistics-based scheme

● Process-based scheme

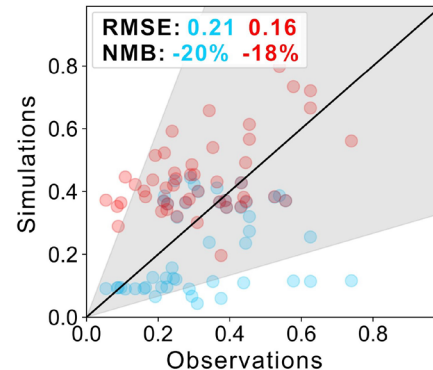
□ Good performance on Hg^0 and Hg^{II} simulations

□ Insignificant difference of the performance between these two schemes

Reproducing Hg^{P} concentrations and F_p in the marine atmosphere



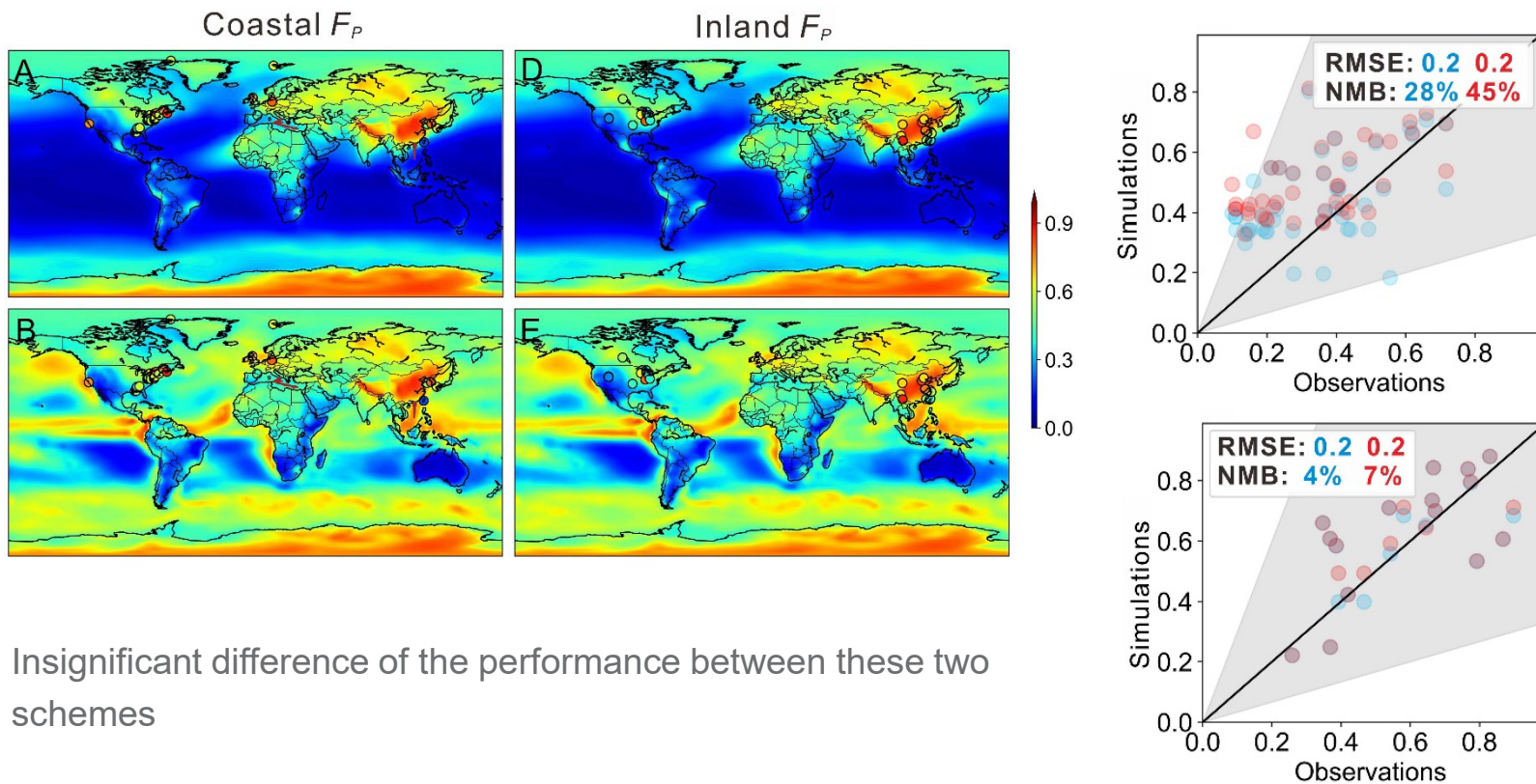
● Statistics-based scheme



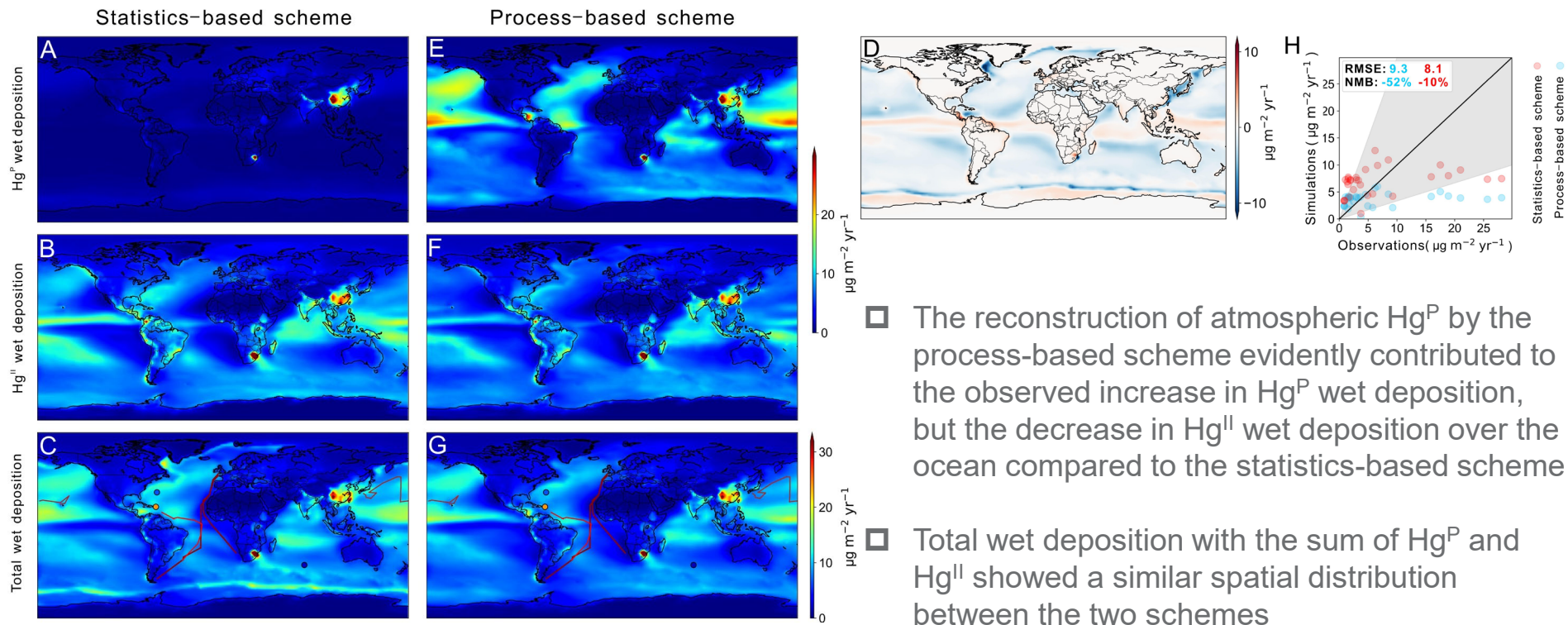
● Process-based scheme

- Significantly improve the model performance on Hg^{P} cycling in the marine atmosphere using the process-based scheme

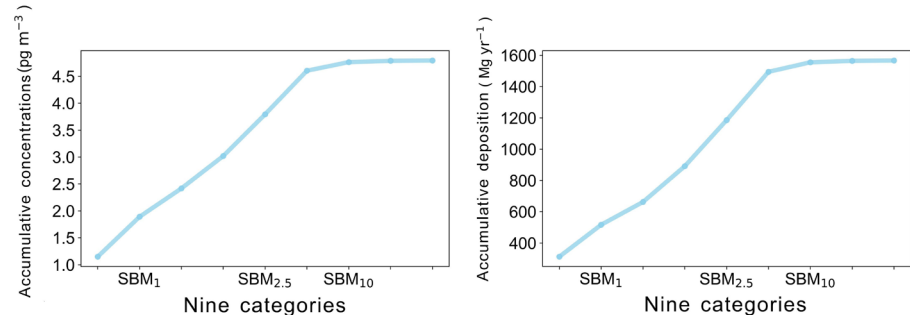
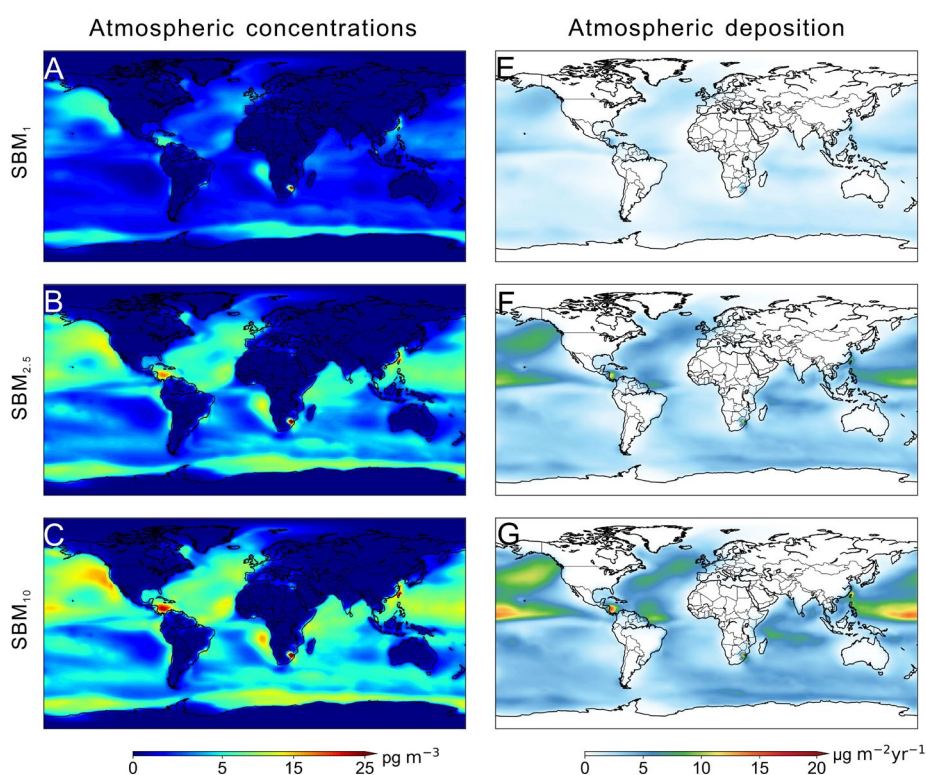
Remaining Hg^p concentrations and F_p in the continental atmosphere



Improved model performance on wet deposition over the ocean

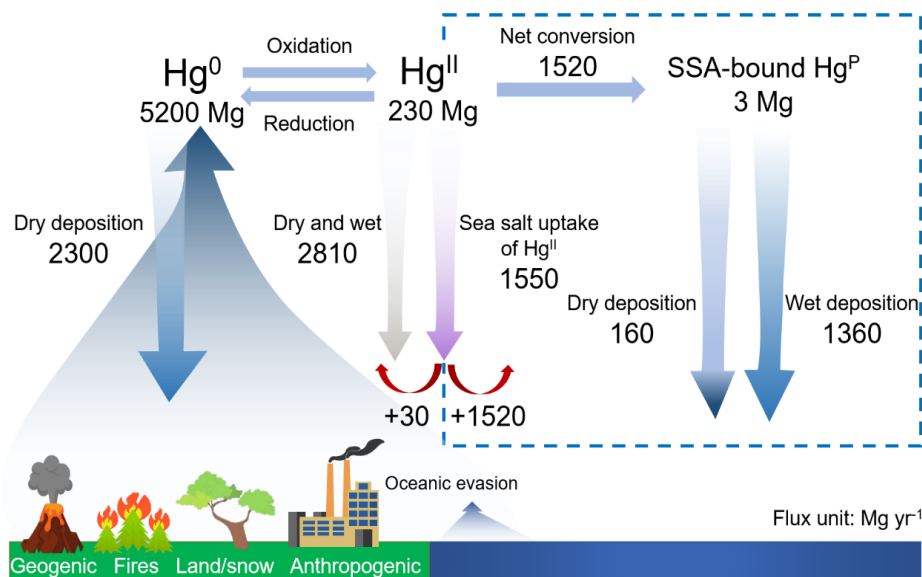


Global distribution of size-resolved sea salt aerosol-bound Hg^p

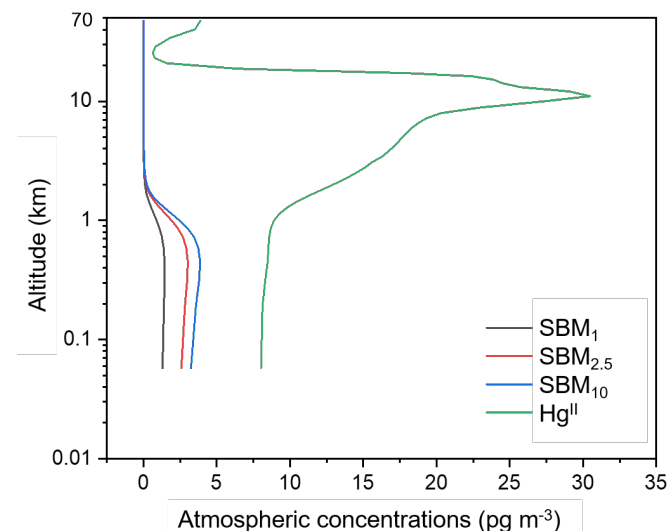


- Atmospheric concentrations and annual deposition flux of SBM₁ were much lower than those of SBM_{2.5} and SBM₁₀, and the magnitudes of SBM_{2.5} and SBM₁₀ were comparable
- Hotspots were observed over Northern Hemisphere waters, such as the North Pacific and North Atlantic
- The global total deposition fluxes of SBM₁, SBM_{2.5}, and SBM₁₀ were estimated to be **515 Mg yr⁻¹**, **1,145 Mg yr⁻¹**, and **1,550 Mg yr⁻¹** respectively, accounting for **33.0%**, **75.7%**, and **99.3%** respectively.

Implications for global atmospheric mercury budget

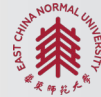


- The major change of the new budget was the **transition** from gaseous Hg^{II} deposition to particulate Hg^{P} deposition over the ocean



- The atmospheric burden of SSA-bound Hg^{P} was low (3 Mg), because the formation of SSA-bound Hg^{P} occurred mainly in air columns with limited height

- ❑ The process-based scheme reproduced Hg^{P} concentrations and F_p in the marine atmosphere
- ❑ A new global atmospheric Hg cycling budget was constructed that accounted for the SSA-bound Hg^{P} cycling in the marine atmosphere. The major change of the new budget was the **transition** from gaseous Hg^{II} deposition to particulate Hg^{P} deposition over the ocean
- ❑ Atmospheric burden of SSA-bound Hg^{P} was low (3 Mg), because the formation of SSA-bound Hg^{P} occurred mainly in air columns with limited height



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THANKS!



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